

Answers to Option C test yourself questions

1 a 70%; b 18%

2 40%

3 a 22.7 kJ g^{-1} ; b 48.3 kJ g^{-1} ; c 33.2 kJ g^{-1}

4 a 35.9 kJ dm^{-3} ; b 52.5 kJ dm^{-3} ; c $39\,600 \text{ kJ dm}^{-3}$;
d $23\,400 \text{ kJ dm}^{-3}$

5 a M; b P; c Q

6 a $\text{C}_{13}\text{H}_{28} \rightarrow \text{C}_8\text{H}_{18} + \text{C}_5\text{H}_{10}$

b $\text{C}_{10}\text{H}_{22} \rightarrow \text{C}_6\text{H}_{14} + \text{C}_4\text{H}_8$

c $\text{C}_{16}\text{H}_{34} \rightarrow \text{C}_9\text{H}_{18} + \text{C}_7\text{H}_{16}$

7 a *Ethanol*: 1.910 g of CO_2 released per gram of ethanol burned

0.06439 g of CO_2 released per kJ of energy released

b *Hexane*: 3.063 g of CO_2 released per gram of hexane burned

0.06343 g of CO_2 released per kJ of energy released

8 a 43.3 kg CO_2e ; b 248 kg CO_2e

9 a ${}^{235}_{92}\text{U} + {}^1_0\text{n} \rightarrow {}^{236}_{92}\text{U} \rightarrow {}^{140}_{55}\text{Cs} + {}^{93}_{37}\text{Rb} + 3{}^1_0\text{n}$

b ${}^{235}_{92}\text{U} + {}^1_0\text{n} \rightarrow {}^{236}_{92}\text{U} \rightarrow {}^{105}_{42}\text{Mo} + {}^{129}_{50}\text{Sn} + 2{}^1_0\text{n}$

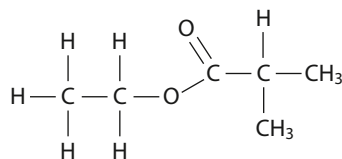
10 a ${}^{121}_{48}\text{Cd}$; b ${}^{96}_{40}\text{Zr}$

11 a 25 mg; b 6.25 mg; c 0.0980 mg

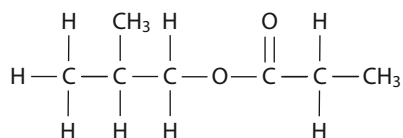
12 a 45 days; b 7×10^{11} y

13 a 800 y; b 168 d

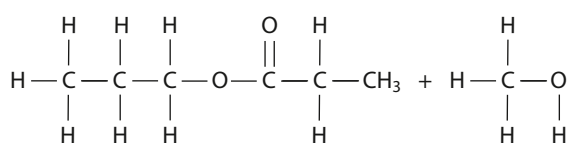
14 a



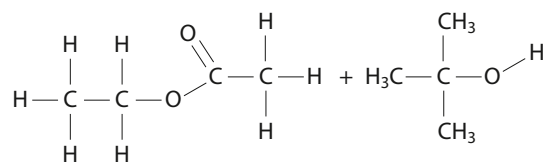
b



15 a



b



16 HCN, HF, CO, CO_2 , H_2S , CFCl_3 , N_2O

17 Three of: carbon dioxide; methane; nitrous oxide [nitrogen(I) oxide]; chlorofluorocarbons

18 a false; b false; c true; d true

19 a $\text{PbSO}_4(\text{s}) + 2\text{e}^- \rightarrow \text{Pb}(\text{s}) + \text{SO}_4^{2-}(\text{aq})$

b $\text{Ni}(\text{OH})_2(\text{s}) + \text{OH}^-(\text{aq}) \rightarrow \text{NiO}(\text{OH})(\text{s}) + \text{H}_2\text{O}(\text{l}) + \text{e}^-$

20 $6\text{C} + \text{Li}_x\text{CoO}_2 \rightarrow \text{Li}_x\text{C}_6 + \text{CoO}_2$

21 a -0.17 V ; b 0.68 V ; c 0.81 V ; d 1.82 V ; e -0.77 V

22 a 1.17 V ; b 1.32 V ; c 1.35 V

23 a 0.39 V ; b 1.07 V ; c 0.58 V

24 a 0.06 V ; b 0.03 V ; c 0.04 V ; d 0.09 V

25 a mass defect 0.069511 u

binding energy $1.04 \times 10^{-11} \text{ J}$

binding energy per nucleon 6.48 MeV

b mass defect 0.20028 u

binding energy $2.99 \times 10^{-11} \text{ J}$

binding energy per nucleon 8.12 MeV

c mass defect 0.52845 u

binding energy $7.90 \times 10^{-11} \text{ J}$

binding energy per nucleon 8.80 MeV

d mass defect 1.9398 u

binding energy $2.90 \times 10^{-10} \text{ J}$

binding energy per nucleon 7.57 MeV

26 a $3.89 \times 10^8 \text{ kJ mol}^{-1}$; b $1.70 \times 10^9 \text{ kJ mol}^{-1}$

27 a $1.68 \times 10^{10} \text{ kJ mol}^{-1}$; b $1.84 \times 10^{10} \text{ kJ mol}^{-1}$

28 a 0.012 min^{-1} ; b $8.04 \times 10^{-3} \text{ d}^{-1}$; c $3.30 \times 10^{-16} \text{ y}^{-1}$

29 a 134 minutes (135 minutes if all figures are carried through on the calculator); b 200 days; c 4.9×10^{15} years

30 a 29.8%; b 16.4%

31 Half-life = $1.33 \times 10^6 \text{ s}$ or 15.4 days

Mass left = $0.301 \mu\text{g}$

32 a helium effuses at 4.00 times the rate of sulfur dioxide

b ethene effuses at 1.25 times the rate of propane

c hydrogen effuses at 2.82 times the rate of methane

- 33 a** 32.1 g mol^{-1} ; **b** 26.0 g mol^{-1} ; **c** 27.0 g mol^{-1}
- 34 a** $3.85 \times 10^{-3} \text{ mol}$ methane and $2.69 \times 10^{-3} \text{ mol}$ helium
- b** $4.14 \times 10^{-3} \text{ mol}$ carbon monoxide and $4.31 \times 10^{-3} \text{ mol}$ carbon dioxide
- c** $3.56 \times 10^{-3} \text{ mol}$ fluorine and $3.94 \times 10^{-3} \text{ mol}$ chlorine
- 35** (shortest wavelength) II < V < IV < I < III < VI
- 36 a** p-type; **b** p-type; **c** n-type; **d** n-type